

**SURFACE TENSION**

- Mercury does not wet glass, wood or iron because
  - Cohesive force is less than adhesive force
  - Cohesive force is greater than adhesive force
  - Angle of contact is less than  $90^\circ$
  - Cohesive force is equal to adhesive force
- The spherical shape of rain-drop is due to
  - Density of the liquid
  - Surface tension
  - Atmospheric pressure
  - Gravity
- At which of the following temperatures, the value of surface tension of water is minimum
  - $4^\circ C$
  - $25^\circ C$
  - $50^\circ C$
  - $75^\circ C$
- The main difference between a stretched membrane and the liquid surface is
  - The liquid surface has a tendency to contract but the stretched membrane does not
  - The surface tension does not depend on area but on the tension of the stretched membrane does
  - The surface tension increases with increases in area
  - Surface tension increases irregularly with temperature
- On bisecting a soap bubble along a diameter, the force due to surface tension on any of its half part will be
  - $4\pi RT$
  - $\frac{4\pi R}{T}$
  - $\frac{T}{4\pi R}$
  - $\frac{2T}{R}$
- 1000 drops of water all of same size join together to form a single drop and the energy released raises the temperature of the drop. Given that  $T$  is the surface tension of water,  $r$  the radius of each small drop,  $\rho$  the density of liquid,  $J$  the mechanical equivalent of heat. What is the rise in the temperature
  - $T/Jr$
  - $10T/Jr$
  - $100T/Jr$
  - None of these
- Two bubbles  $A$  and  $B$  ( $A > B$ ) are joined through a narrow tube. Then
  - The size of  $A$  will increase
  - The size of  $B$  will increase
  - The size of  $B$  will increase until the pressure equals
  - None of these
- Excess pressure of one soap bubble is four times more than the other. Then the ratio of volume of first bubble to another one is
  - 1 : 64
  - 1 : 4
  - 64 : 1
  - 1 : 2
- The angle of contact between glass and mercury is
  - $0^\circ$
  - $30^\circ$
  - $90^\circ$
  - $135^\circ$
- When the temperature is increased the angle of contact of a liquid
  - Increases
  - Decreases
  - Remains the same
  - First increases and then decreases
- Water rises to a height of  $10\text{cm}$  in capillary tube and mercury falls to a depth of  $3.112\text{cm}$  in the same capillary tube. If the density of mercury is 13.6 and the angle of contact for mercury is  $135^\circ$ , the ratio of surface tension of water and mercury is
  - 1 : 0.15
  - 1 : 3
  - 1 : 6
  - 1.5 : 1
- Water can rise to a height  $h$  in a capillary tube lowered vertically into water. If the height of tube above the surface of water be  $l$  and  $l < h$ , then water will rise in the capillary to a height
  - $h$
  - $l$
  - $l - h$
  - $l + h$
- Water rises in a capillary tube through a height  $h$ . If the tube is inclined to the liquid surface at  $30^\circ$ , the liquid will rise in the tube upto its length equal to
  - $\frac{h}{2}$
  - $h$
  - $2h$
  - $4h$
- In the above question, the radius of the bigger drop will be
  - $\sqrt{3}\text{ cm}$
  - $\sqrt{5}\text{ cm}$
  - $\sqrt{7}\text{ cm}$
  - $\sqrt{8}\text{ cm}$
- The surface tension for pure water in a capillary tube experiment is
  - $\frac{\rho g}{2hr}$
  - $\frac{2}{hr\rho g}$
  - $\frac{r\rho g}{2h}$
  - $\frac{hr\rho g}{2}$